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following molecular shapes: Trigonal planar AX E Bent  $< 109.50$  Notation Geometry Electron group arrangement Molecular Shape Idealized Bond An les Linear 1800 A)ÇE Bent (V-shaped)  $< 1200$  Trigonal pyramidal  $< 109.50$  Tetrahedral 109.50 17. Draw the Lewis Dot structure of 1-12 IL) 18.

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Molecular Geometry How can molecular shapes be predicted using the VSEPR theory? Why? When you draw a Lewis structure for a molecule on paper, you are making a two-dimensional representation of the atoms. In reality however, molecules are not flat—they are three-dimensional. The true shape of a molecule is important because it determines many physical and chemical properties for the substance.

## **can shapes be predicted by the theory? Why? - LTHS Answers**

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Other Results for Molecular Geometry

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Worksheet Answer Key:

[www.hudson.k12.oh.us](http://www.hudson.k12.oh.us). Extension - VSEPR Worksheet #2 For this activity, you will need to refer to your Lewis Dot Worksheet # 1. Use the ball and stick model kit to create the molecules below. Then, draw a ball and stick model to show the 3-D structure for each molecule.

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structures shapes and polarity.  
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Key This total will be equal to a value  
1-6.

## **Worksheet 15 Molecular Shapes Answers**

Molecule Polarity Phet Lab Answer Key  
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Polarity Lab Introduction: In this atomic-  
level simulation, you will investigate how  
atoms' electronegativity value Page 2/7.  
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## **Molecular Polarity Lab Answer Key**

13, Based on the information in the Read  
This! section, sketch one of the  
molecular shapes shown below in each  
of the boxes provided in Model I.  
Trigonal planar 120° Bent 104.5° Three-  
Dimensional Molecular Shapes  
Tetrahedral 109.5° STOP Molecular

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Geometry Linear Pyramidal 1070

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