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Ex. 1 dozen = 12 Chapter 10 10.1 The Mole: A Measurement of Matter Chemist use a unit that is a specified unit of particles. This unit is called the mole.

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At STP, one mole, or 6.02×10^{23} particles of any gas occupies a volume of 22.4 L or 22.4 dm^3 , used to convert between the number of moles of gas and the volume of gas at STP $1 \text{ L} = 1 \text{ dm}^3$

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