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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

5 The Periodic Law

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Chemistry: Chapter 5 Review Describe... 1) Dalton's model - solid sphere, made up of only 1 type of matter, no space 2) Schrodinger's model - based on probability of finding an e-; electrons are found in orbitals (look like electron clouds), which are in sublevels (s,p,d,f,g), which are in

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Identify the letter of the choice that best completes the statement or answers the question. ____ 1. The idea of arranging the elements in the periodic table according to their chemical and physical properties is attributed to. a. Mendeleev. c. ... Chemistry Chapter 5 Exam ...

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A 3.00 liter mixture of CO₂ and O₂ gas, total pressure 740 Torr and temperature 50°C, is scrubbed with a water spray at 20°C so as to dissolve all the CO₂ and remove it from the gas phase. If the final observed gas volume at 740 Torr and 20°C is 2.50 liters, what percent of the molecules in the original mixture must have been CO₂? Assume that the vapor pressure of water at 20°C is 17.55 ...

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Assessment Chapter Test A - Wag & Paws

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